

AN EFFICIENT SYNTHESIS OF A KEY INTERMEDIATE FOR
OPTICALLY ACTIVE 5,6-CIS-CARBAPENEM ANTIBIOTICS

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Summary: A versatile intermediate (**18**) for optically active 5,6-cis-carbapenem antibiotics was synthesized with a highly regioselective intramolecular aldol condensation as a key step.

Since the discovery of 5,6-cis-carbapenems, represented by C-19393 H₂¹ (carpetimycin A²), they have attracted much attention as synthetic targets because of their potent and broad antibacterial activity. Several synthetic methods of constructing 5,6-cis-carbapenems have been reported.³

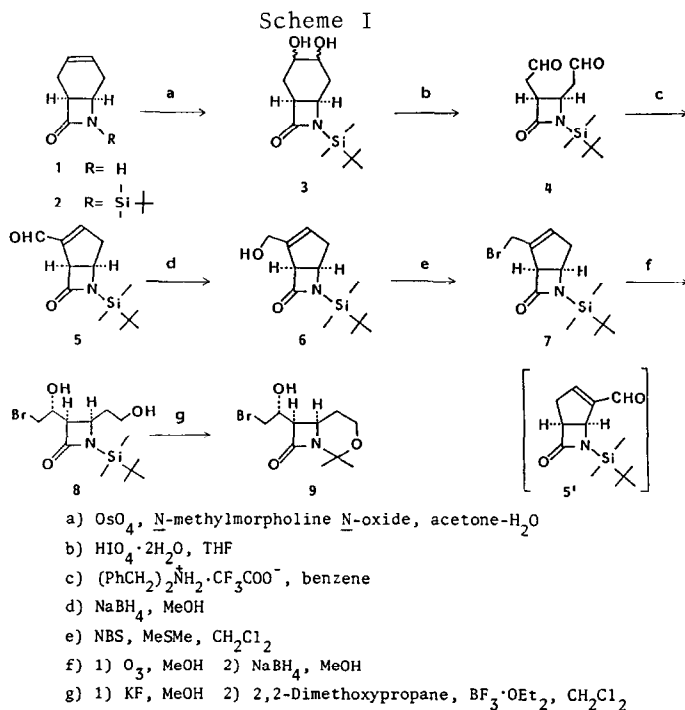
In the preceding papers,⁴ we described a stereoselective synthesis of racemic 5,6-cis-carbapenems via reductive desulfurization utilizing an organotin hydride. Since then our efforts have been concentrated on preparing a chiral 5,6-cis-carbapenem with a functional group, in addition to a hydroxy group, in the C-6 side chain to obtain new carbapenem antibiotics with improved biological activity. Here we report a novel stereocontrolled synthesis of **18** which may serve as a potential intermediate for this purpose.

The starting material selected for our initial investigation (shown in Scheme I) was easily available 8-oxo-7-azabicyclo[4.2.0]oct-2-ene (**1**).⁵ **1** was treated with tert-butyldimethylsilyl chloride-Et₃N and then with N-methylmorpholine N-oxide and osmium tetroxide⁶ to afford cis-diol **3** in 63 % yield. Treatment of **3** with HIO₄·2H₂O gave dialdehyde **4**. Reaction conditions were examined to convert **4** into **5** selectively. First, the rather unstable dialdehyde **4** was treated with morpholine-camphoric acid⁷ to give ca. 1:1 mixture of **5** and **5'**. Then, by employing piperidine-acetic acid,⁸ **5** was obtained highly regioselectively but in a low yield (after reduction with NaBH₄, **6**⁹ was isolated in 16 % yield). But when the dialdehyde **4** was treated with dibenzylammonium trifluoroacetate,¹⁰ the intramolecular aldol condensation proceeded highly regioselectively and gave **5** exclusively in a good yield (**5'** was not observed by 90 MHz NMR and by HPLC). We presume that this high selectivity might be due to the bulkiness of the tert-butyldimethylsilyl group and the amine component. Unstable α,β -unsaturated aldehyde **5** was reduced with NaBH₄ to **6** without isolation (57 % yield from **3**). **6** was converted into the

bromide **7** with *N*-bromosuccinimide-dimethyl sulfide¹¹ in 67 % yield. **7** was subjected to ozonolysis, then, to reduction with NaBH₄ to afford diol **8** in 97 % yield (for the stereochemistry vide post). **8** was deprotected with potassium fluoride and then treated with 2,2-dimethoxypropane (BF₃·OEt₂, CH₂Cl₂) to give acetonide **9**¹² in 59 % yield.

This sequence of reactions was applied to the optically active compound (**17**) to obtain **18** with optical activity. **17** was prepared from the racemic compound **10** by successfully combining optical resolution and effective conversion of both enantiomers (**11a**¹³ and **11b**^{13b}) into the desired optically active compound **13** (Scheme II). Thus, resolution of **10** was effected with cinchonidine to give chiral **11a** [28 %, [α]_D²⁵+3.36°(c=1.24, CHCl₃)] and its enantiomer **11b** [27 %, [α]_D²⁵-3.44°(c=1.685, CHCl₃)]. **11a** was converted into **12** by Curtius rearrangement in 79 % yield, then **12** was hydrolyzed to **13**[[α]_D²⁵+36.4°(c=0.45, H₂O)] in 76 % yield. **13** was transformed to the optically active β-lactam **17**¹⁴ [mp 163-164°C, [α]_D²⁵-28.6°(c=0.585, CHCl₃)] with Ph₃P-(PyS)₂/CH₃CN.¹⁵ On the other hand **11b** was also converted into **13** [[α]_D²⁵+36.6°(c=0.56, H₂O)] via esterification (92 %), hydrolysis (93 %), Curtius rearrangement (96 %), and deprotection (73 %). The merit of this method is that both the enantiomers (**11a** and **11b**) resolved from **10** are utilized in preparing the optically active compound **17**.

Application of the same sequence of reactions employed in the preparation of **9** from **1** to the optically active compound **17** gave the desired optically active compound **18** [mp 141-142°C, [α]_D²⁵+13.0°(c=0.54, CHCl₃)].



The stereochemistry of **18** was confirmed to be $6R$, $7R$, $9R$ by a single crystal X-ray analysis¹⁶ (Fig. I). The reduction of ozonide of chiral **7** with NaBH_4 was proved to proceed selectively to give $9R$ stereochemistry. Since the bromo atom in the side chain is susceptible to nucleophiles (including H^-), compound **18** serves as an efficient intermediate for preparing new type of carbapenems (**19**) with a variety of functional groups in the C-6 side chain (Scheme III). The synthesis and biological properties of such 5,6-*cis*-carbapenems will be reported in due course.

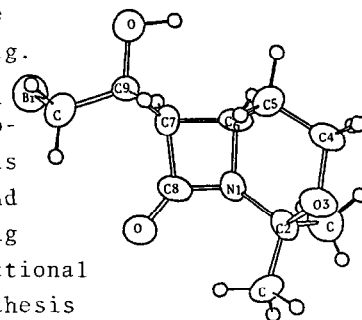
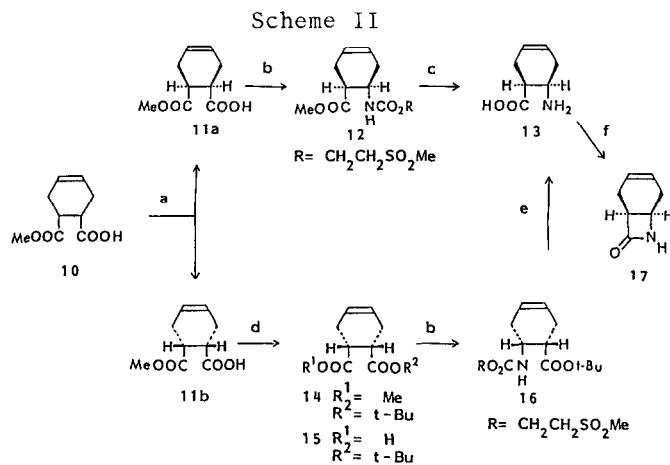
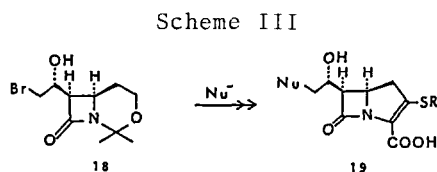


Figure I
ORTEP view of **18**



- a) Optical resolution with cinchonidine, Me_2CO
 b) 1) ClCOEt , Et_3N 2) NaN_3 , $n\text{-Bu}_4\text{NHSO}_4$, CH_2Cl_2 3) 100°C , toluene 4) $\text{MeSO}_2\text{CH}_2\text{CH}_2\text{OH}$
 c) aq. NaOH
 d) 1) isobutene, H_2SO_4 , CH_2Cl_2 2) aq. NaOH
 e) 1) aq. NaOH 2) CF_3COOH
 f) PySSPy , Ph_3P , CH_3CN



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References and notes

- 1) A. Imada, Y. Nozaki, K. Kintaka, K. Okonogi, K. Kitano, and S. Harada, *J. Antibiot.*, 33, 1417 (1980).
- 2) M. Nakayama, A. Iwasaki, S. Kimura, T. Mizoguchi, S. Tanabe, A. Murakami, M. Okuchi, H. Itoh, Y. Saino, F. Kobayashi, and T. Mori, *J. Antibiot.*, 33, 1388 (1980).
- 3) (a) T. Kametani, S.-P. Huang, T. Nagahara, and M. Ihara, *J. Chem. Soc., Perkin Trans. I*, 1981, 2282 ; (b) J. H. Bateson, R. I. Hickling, P. M. Roberts, T. C. Smale, and R. Southgate, *J. Chem. Soc., Chem. Commun.*, 1980, 1084 ; (c) M. Ihara, F. Konno, K. Fukumoto, and T. Kametani, *Heterocycles*, 20, 2181 (1983) ; (d) T. Iimori, Y. Takahashi, T. Izawa, S. Kobayashi, and M. Ohno, *J. Am. Chem. Soc.*, 105, 1659 (1983) ; (e) M. Aratani, H. Hirai, K. Sawada, A. Yamada, and M. Hashimoto, *Tetrahedron Lett.*, 26, 223 (1985) ; (f) A. Knierzinger and A. Vasella, *J. Chem. Soc., Chem. Commun.*, 1984, 9.
- 4) (a) H. Natsugari, Y. Matsushita, N. Tamura, K. Yoshioka, and M. Ochiai, *J. Chem. Soc., Perkin Trans. I*, 1983, 403 ; (b) H. Natsugari, Y. Matsushita, N. Tamura, K. Yoshioka, M. Kondo, K. Okonogi, M. Kuno, and M. Ochiai, *J. Antibiot.*, 36, 855 (1983).
- 5) L. A. Paquette and T. Kakihana, *J. Am. Chem. Soc.*, 90, 3897 (1968).
- 6) V. VanRheenen, R. C. Kelly, and D. Y. Cha, *Tetrahedron Lett.*, 1976, 1973.
- 7) T. Harayama, M. Takatani, and Y. Inubushi, *Chem. Pharm. Bull.*, 28, 1276 (1980).
- 8) R. B. Woodward, F. Sondheimer, D. Taub, K. Heusler, and W. M. McLamore, *J. Am. Chem. Soc.*, 74, 4223 (1952).
- 9) ^{13}C -NMR (CDCl_3) δ : -5.17 (s, SiMe), -5.59 (s, SiMe), 26.23 (q, 3xMe), 37.27 (t, C_4), 52.52 (d, C_5), 60.73 (t, CH_2O), 63.99 (d, C_3), 140.72 (s, C_2), 175.79 (s, C_7).
- 10) E. J. Corey, R. L. Danheiser, S. Chandrasekaran, P. Siret, G. E. Keck, and J.-L. Gras, *J. Am. Chem. Soc.*, 100, 8031 (1978).
- 11) E. J. Corey, C. U. Kim, and M. Takeda, *Tetrahedron Lett.*, 1972, 4339.
- 12) mp 133-135°C, IR $\nu_{\text{max}}^{\text{KBr}}$: 1740 cm^{-1} , ^1H -NMR (CDCl_3) δ : 1.40, 1.69 (each 3H, 2xs, 2xMe), 1.7-2.0 (2H, m, $\text{C}_4\text{-H}_2$), 2.5 (1H, br, OH), 3.29 (1H, dd, J=11, 6Hz, $\text{C}_7\text{-H}$), 3.5-4.3 (6H, m, $\text{C}_4\text{-H}_2$, $\text{C}_6\text{-H}$, CH-O, CH_2Br).
- 13) (a) M. Schneider, N. Engel, P. Hönicke, G. Heineman, and H. Görisch, *Angew. Chem. Int. Ed. Engl.*, 23, 67 (1984) ; (b) H.-J. Gais and K. L. Lukas, *ibid.*, 23, 142 (1984).
- 14) After completion of this work, Ohno reported the synthesis of 17 from 11a which was obtained by enzymatic procedure, see: M. Kurihara, K. Kamiyama, S. Kobayashi, and M. Ohno, *Tetrahedron Lett.*, 26, 5831 (1985).
- 15) S. Kobayashi, T. Iimori, T. Izawa, and M. Ohno, *J. Am. Chem. Soc.*, 103, 2406 (1981).
- 16) We thank Dr. K. Kamiya and Mr. Y. Wada of this Division for this analysis.